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#### **Key indicators**

Single-crystal X-ray study T = 294 K Mean  $\sigma$ (C–C) = 0.006 Å R factor = 0.050 wR factor = 0.137 Data-to-parameter ratio = 12.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 5-Formyl-2-methoxyphenyl benzenesulfonate

In the title compound,  $C_{14}H_{12}O_5S$ , the isovanillin group makes a dihedral angle of 37.45 (15)° with the phenyl ring. The crystal packing is stabilized by a weak non-classical intermolecular  $C-H \cdots O$  hydrogen bond that links molecules into a chain. Received 31 October 2006 Accepted 16 November 2006

## Comment

Schiff base ligands have received a good deal of attention in biology and chemistry (Kahwa *et al.*, 1986; Larson & Pecoraro, 1991). Many Schiff base derivatives have been synthesized and employed to develop protein and enzyme mimics (Santos *et al.*, 2001). As part of our interest in the coordination properties of Schiff bases functioning as ligands (Zhang *et al.*, 2006), we report here the molecular structure of the title compound, (I), which is used as a precursor in the preparation of Schiff bases.



Bond lengths and angles in (I) (Fig. 1) are within normal ranges (Allen *et al.*, 1987). The isovanillin group (atoms C7–C13/O3/O4) is essentially planar, with an r.m.s. deviation for fitted atoms of 0.0075 Å. This group makes a dihedral angle of 37.45 (15)° with the phenyl ring (C1–C6). The crystal packing is stabilized by a weak non-classical intermolecular C– $H \cdots O$ —C hydrogen bond (Table 1) that links adjacent molecules into a chain running along the *c* axis (Fig. 2).



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# The molecular structure of (I), with displacement ellipsoids for non-H atoms drawn at the 30% probability level.

# organic papers



### Figure 2

A partial packing diagram of (I), viewed along the *a* axis, with hydrogen bonds shown as dashed lines.

### **Experimental**

An anhydrous benzene solution (100 ml) of 3-hydroxy-4-methoxybenzaldehyde (1.52 g, 10 mmol) was added dropwise to a benzene solution (100 ml) of benzenesulfonyl chloride (1.76 g, 10 mmol) and pyridine (0.79 g, 10 mmol) and the mixture was refluxed for 24 h under nitrogen. The solvent was removed and the resultant mixture poured into ice–water (100 ml). The white precipitate was isolated, recrystallized from acetonitrile and then dried in a vacuum to give pure compound (I) in 58% yield. Colorless single crystals suitable for X-ray analysis were obtained by slow evaporation of an acetonitrile solution.

### Crystal data

 $C_{14}H_{12}O_5S$   $M_r = 292.31$ Triclinic,  $P\overline{1}$  a = 7.915 (7) Å b = 8.255 (8) Å c = 11.581 (10) Å  $\alpha = 102.684 (14)^{\circ}$   $\beta = 109.208 (13)^{\circ}$   $\gamma = 97.137 (13)^{\circ}$ 

Data collection

Bruker SMART APEX CCD areadetector diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)  $T_{\min} = 0.918, T_{\max} = 0.960$   $V = 681.1 (11) Å^{3}$  Z = 2  $D_{x} = 1.425 \text{ Mg m}^{-3}$ Mo K\alpha radiation  $\mu = 0.25 \text{ mm}^{-1}$  T = 294 (2) KBlock, colorless  $0.28 \times 0.26 \times 0.16 \text{ mm}$ 

3291 measured reflections 2365 independent reflections 1404 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.041$  $\theta_{\text{max}} = 25.0^{\circ}$  Refinement

$w = 1/[\sigma^2(F_o^2) + (0.066P)^2]$
where $P = (P_o + 2F_c)/3$ $(\Delta/\sigma)_{\text{max}} = 0.004$
$\Delta \rho_{\rm max} = 0.24 \text{ e } \text{\AA}^{-3}$
$\Delta \rho_{\rm min} = -0.40 \text{ e } \text{\AA}^{-3}$
Extinction correction: SHELXL97
Extinction coefficient: 0.064 (8)

**Table 1** Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$C5-H5\cdots O5^i$	0.93	2.56	3.405 (6)	151

Symmetry code: (i) x, y, z + 1.

H atoms were included in calculated positions (C-H = 0.93–0.96 Å) and refined as riding, with  $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C})$  or  $1.5U_{\rm eq}({\rm methyl C})$ .

Data collection: *SMART* (Bruker, 1999); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997*a*); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997*a*); molecular graphics: *SHELXTL* (Sheldrick, 1997*b*); software used to prepare material for publication: *SHELXTL*.

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